

Fig. 1

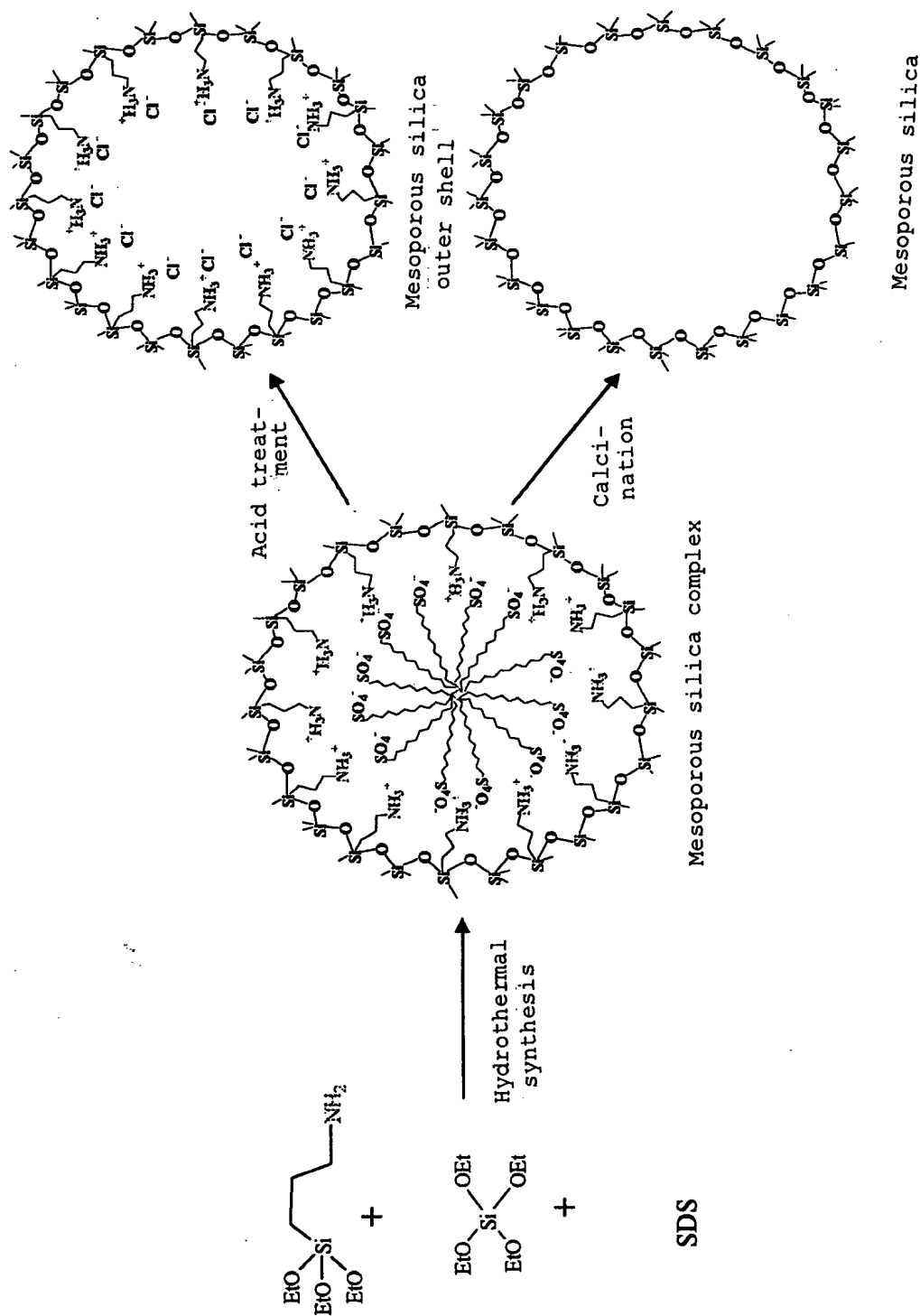


Fig. 2

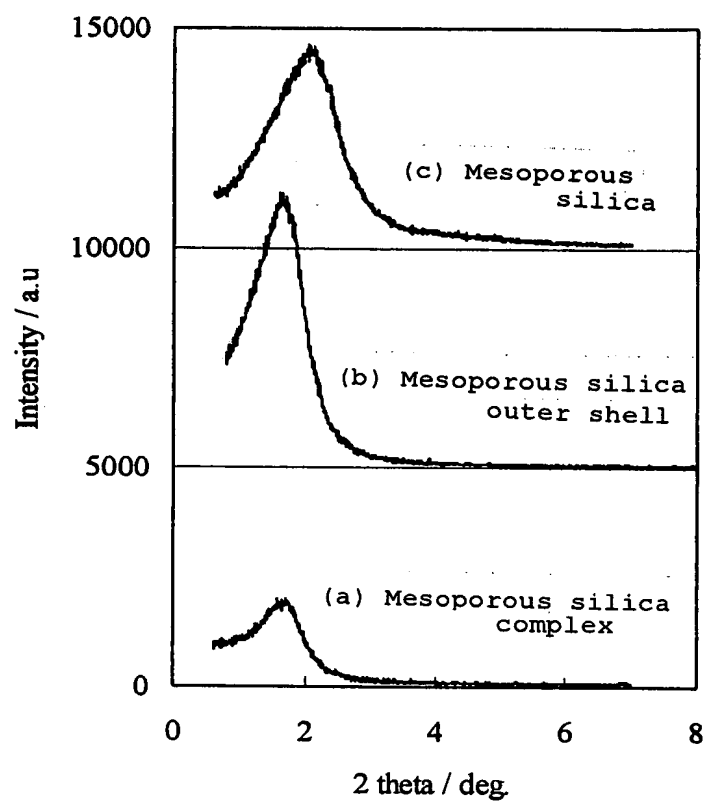
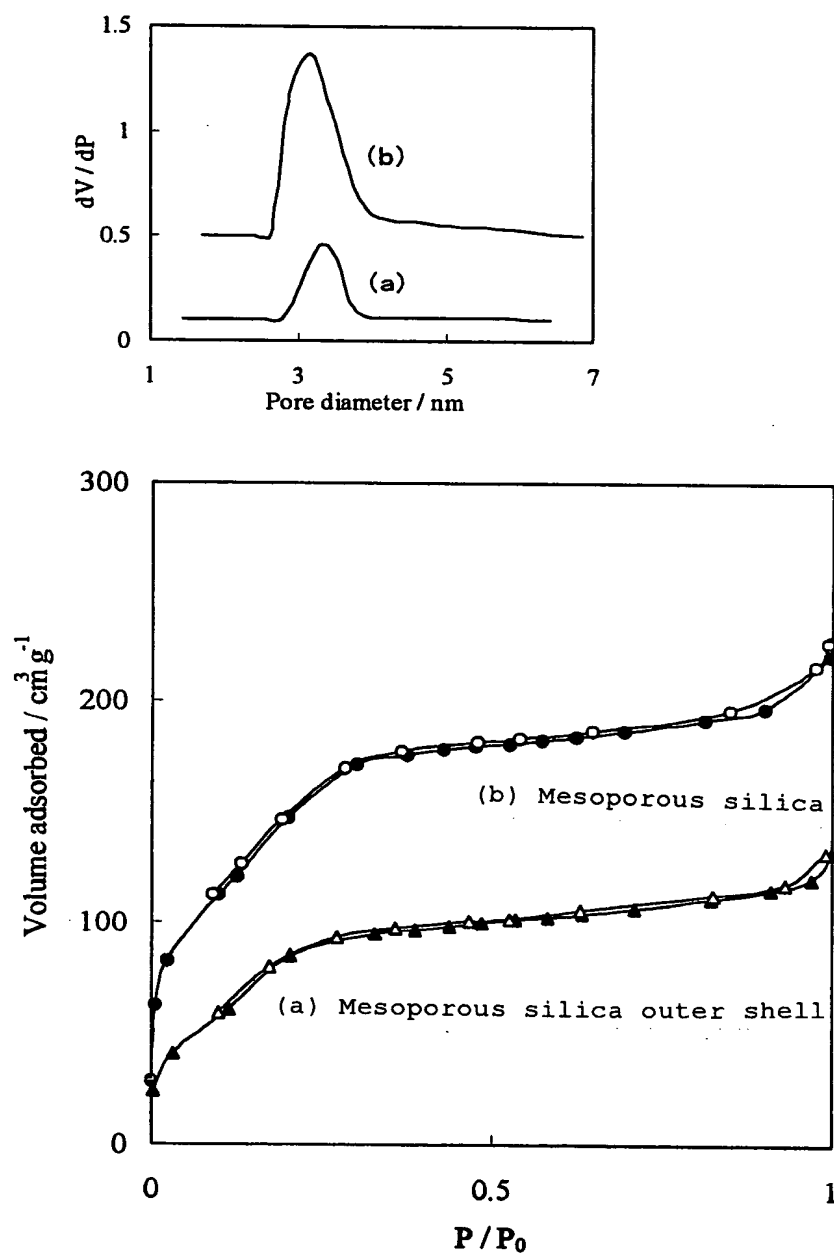


Fig. 3



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Fig. 4

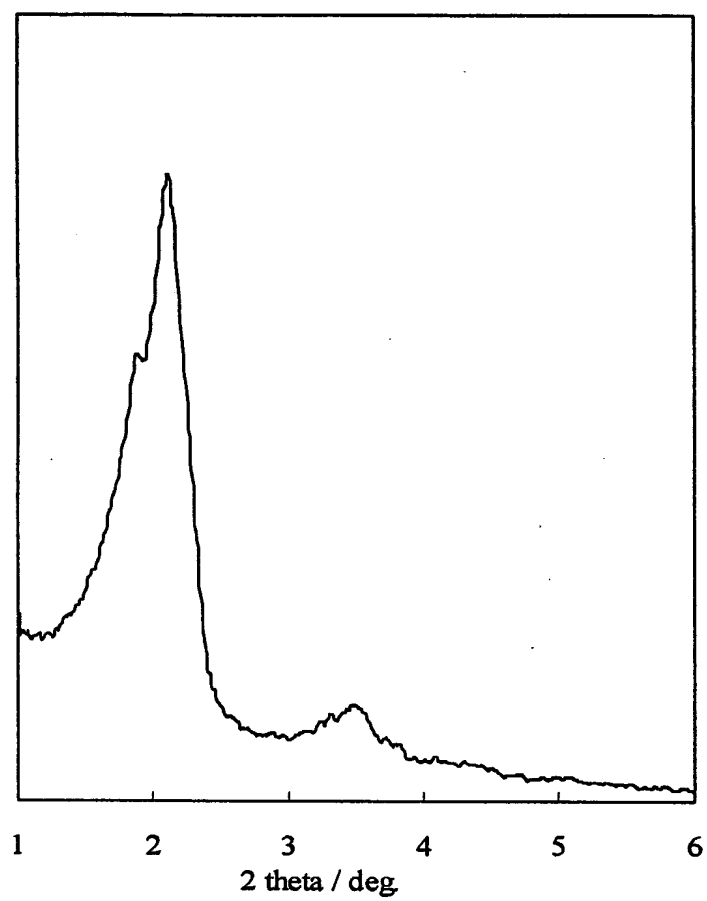


Fig. 5

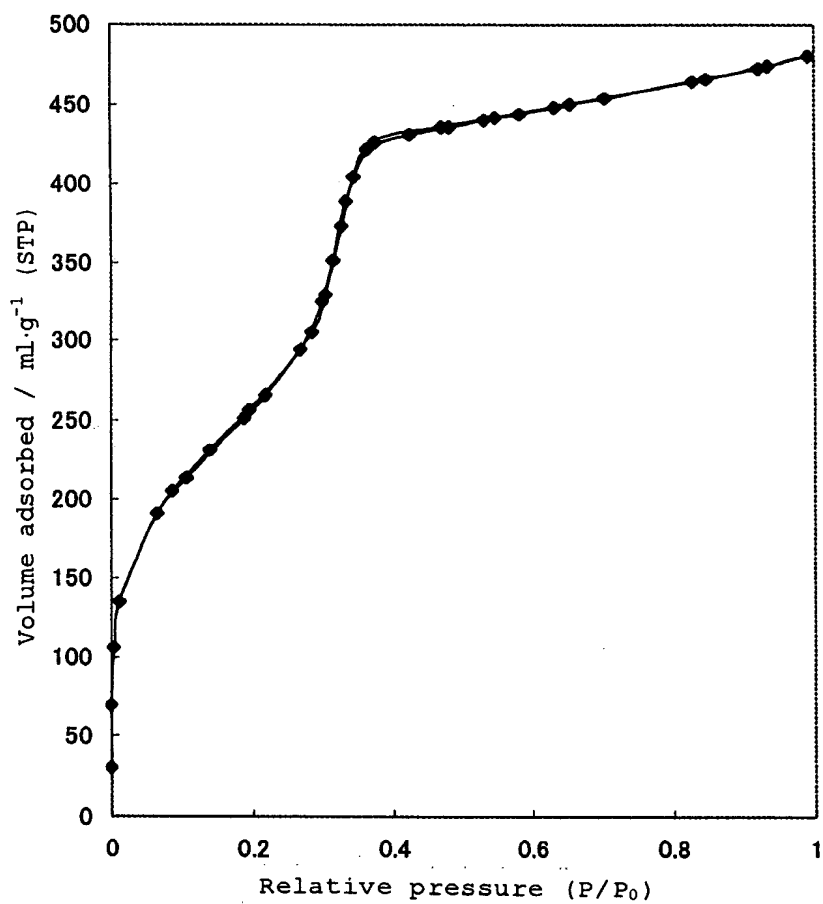
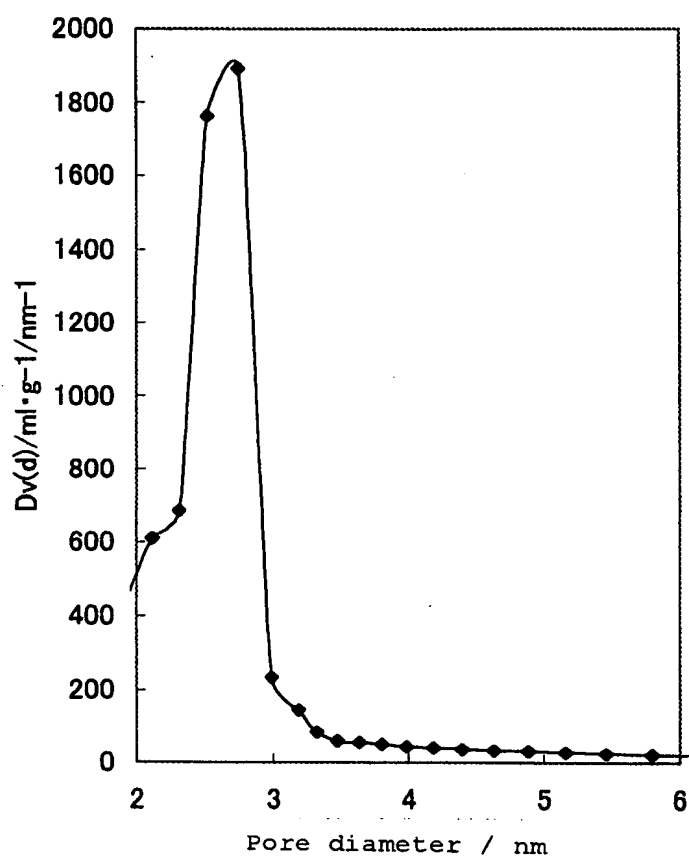


Fig. 6



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Fig. 7

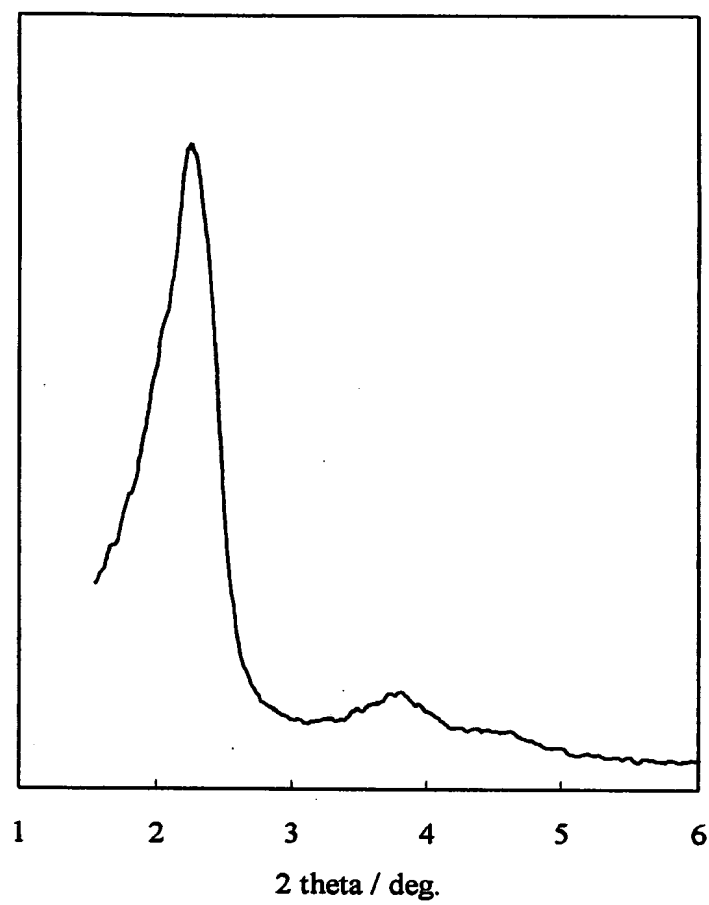


Fig. 8

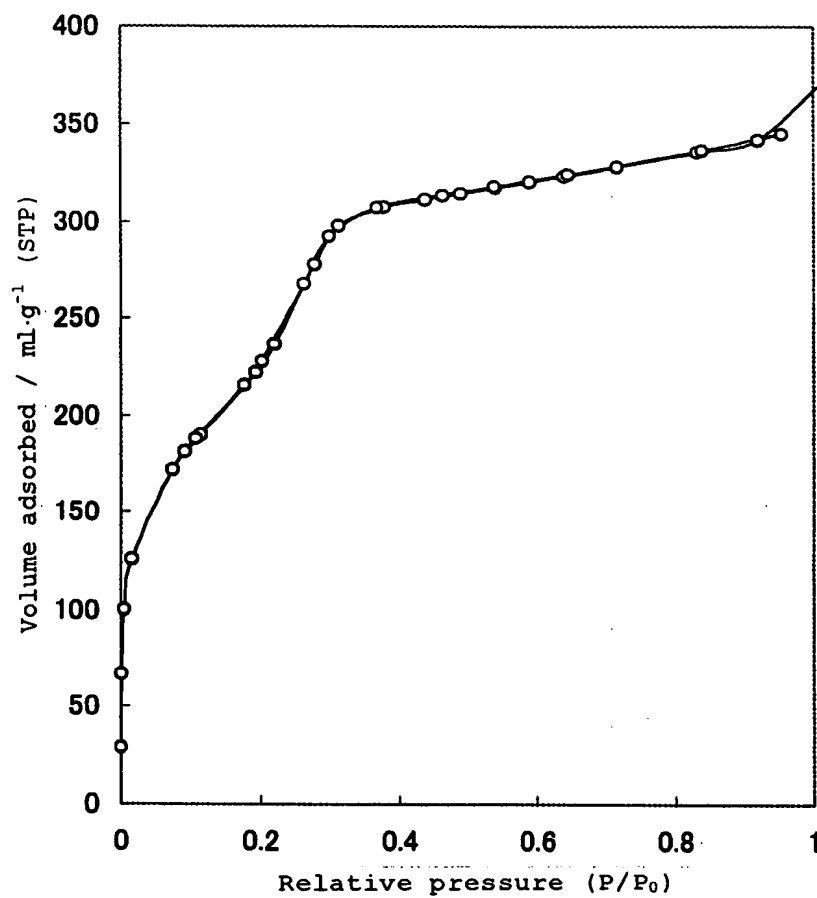
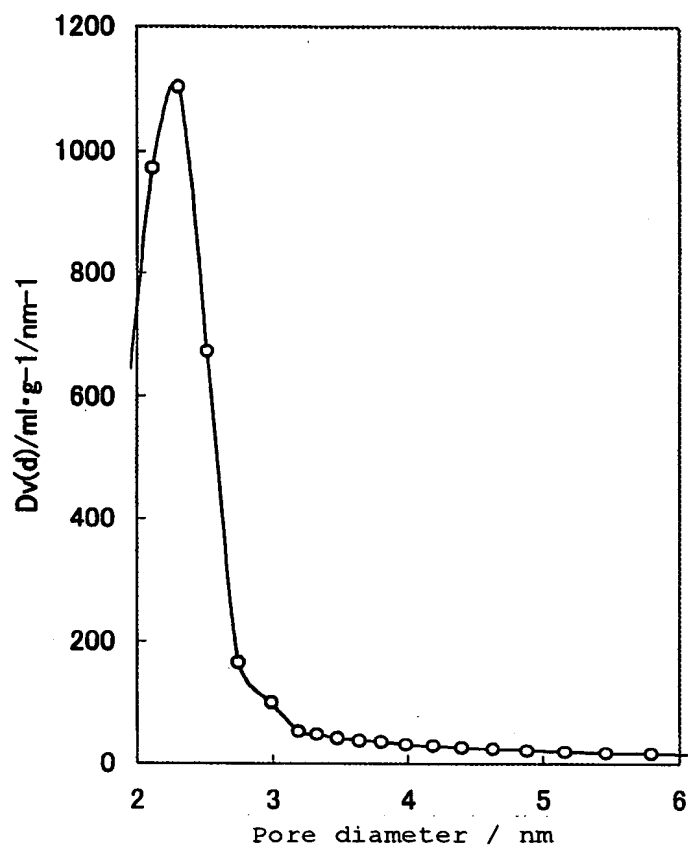


Fig. 9



	Neutralization	Double decomposition
Surfactant	~~~~~AH	$\text{~~~~~A}^- \text{M}^+$
CSDA	$\text{H}_2\text{N}-\text{CH}_2-\text{CH}_2-\text{Si}(\text{OCH}_3)_3$ <p>(APS)</p>	$\text{H}_3\text{C}-\text{N}^+(\text{CH}_3)_3-\text{CH}_2-\text{CH}_2-\text{Si}(\text{OCH}_3)_3 \text{Cl}^-$ <p>(TMAPS)</p>
Interaction		
	$\text{~~~~~} : \text{C}_n\text{H}_{2n+1}, \text{C}_n\text{H}_{2n+1}-\text{C}(=\text{O})-\text{NH}-\underset{\text{R}_1}{\text{CH}}-, \text{C}_n\text{H}_{2n+1}-\text{C}(=\text{O})-\text{NH}-\underset{\text{AH}}{\text{CH}}-$ <p>A: COO, OSO₃, SO₃, OPO₃; M⁺: Na⁺, K⁺, NH₄⁺ etc.; R₁: H, CH₃; n = 8 - 18;</p>	

Fig. 10. Schematic illustration of the two types of amino group-anionic surfactant head group interactions: through neutralization of acid with primary aminosilane APS and double decomposition of negatively charged anionic salt surfactant with positively charged quaternized aminosilane TMAPS.

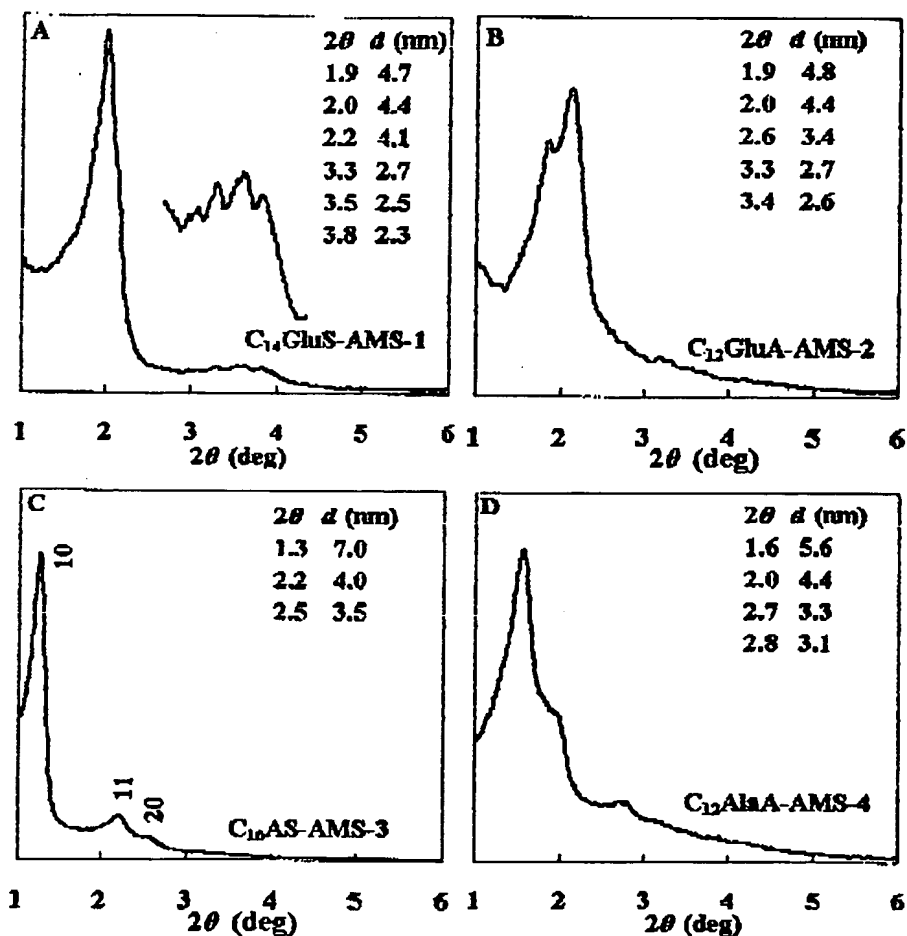


Fig. 11. XRD patterns of calcined AMS-*n* mesoporous silica. The chemical mol composition of the reaction mixture was (A) $C_{14}\text{GluS-AMS-1}$, $C_{14}\text{GluS:TMAPS:TEOS:H}_2\text{O}$ 1:2:10:2405 (at 100 °C for 3 d); (B) $C_{12}\text{GluA-AMS-2}$: $C_{12}\text{GluA:APS:TEOS:H}_2\text{O}$ 1:2.5:18.5:1905 (at 100 °C for 2 d); (C) $C_{16}\text{AS-AMS-3}$: $C_{16}\text{AS:TMAPS:TEOS:H}_2\text{O}$ 1:1:9:1544 (at 60 °C for 1 d); (D) $C_{12}\text{AlaA-AMS-4}$, $C_{12}\text{AlaA:APS:TEOS:H}_2\text{O}$ 1:0.75:7.5:1505 (at 60 °C for 1 d). XRD patterns were recorded on an MX Labo powder diffractometer equipped with Cu K α radiation (40

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kV, 20 mA) at the rate of 1.0 deg/min over the range of 1.5 – 10.0 ° (2θ).

Supporting on line materials:

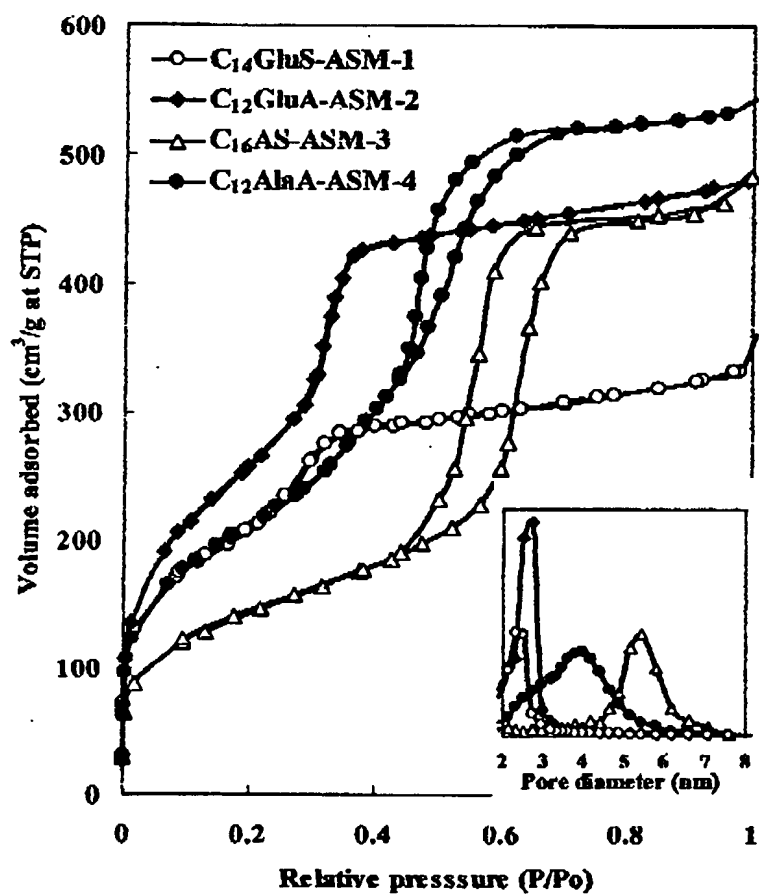


Fig. 12. N₂ adsorption-desorption isotherms and BJH pore size distributions of AMS-n mesoporous silica shown in Fig. 11. The isotherms were measured at -196 °C on a Belsorp 28SA sorptionmeter.

DF4392/US/C-181

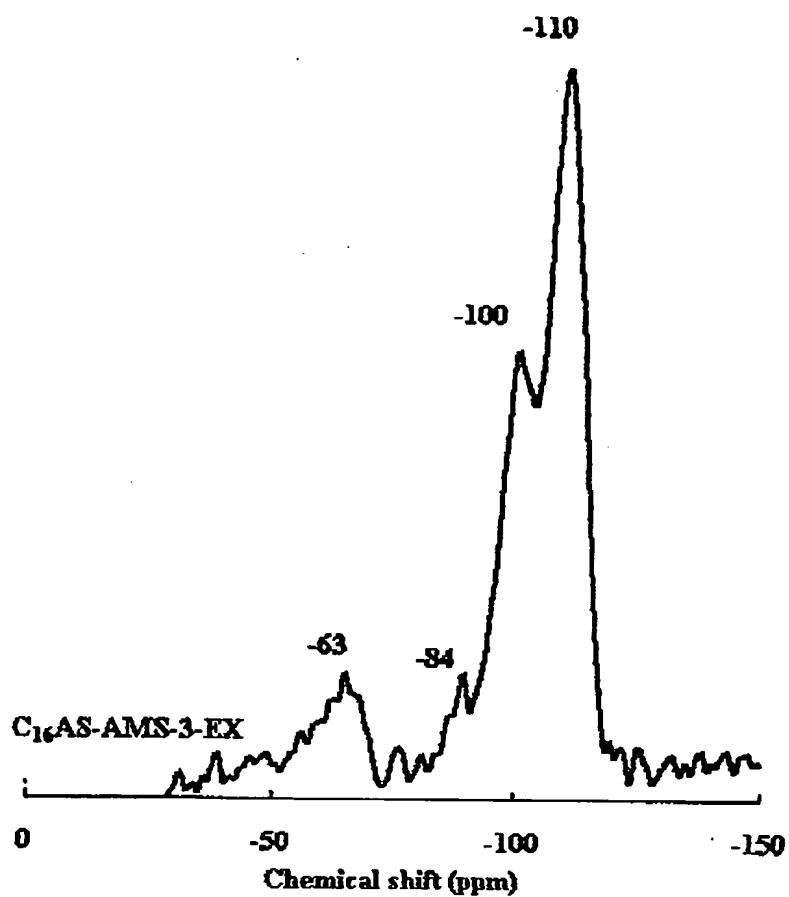


Fig. 13 shows CP ^{29}Si NMR spectra of extracted AMS-3 silica C₁₆AS-AMS-3-Ex. The spectra were collected at a JEOL-LA400WB 400 MHz spectrometer at 79.4 MHz and a sample spinning frequency of 5 kHz, respectively.